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# Greenhouse gas emissions from membrane bioreactors: analysis of a two-year survey on different MBR configurations

Giorgio Mannina, Kartik Chandran, Marco Capodici, Alida Cosenza, Daniele Di Trapani and Mark C. M. van Loosdrecht

## ABSTRACT

This study aimed at evaluating the nitrous oxide (N<sub>2</sub>O) emissions from membrane bioreactors (MBRs) for wastewater treatment. The study investigated the N<sub>2</sub>O emissions considering multiple influential factors over a two-year period: (i) different MBR based process configurations; (ii) wastewater composition (municipal or industrial); (iii) operational conditions (i.e. sludge retention time, carbon-to-nitrogen ratio, C/N, hydraulic retention time); (iv) membrane modules. Among the overall analysed configurations, the highest N<sub>2</sub>O emission occurred from the aerated reactors. The treatment of industrial wastewater, contaminated with salt and hydrocarbons, provided the highest N<sub>2</sub>O emission factor (EF): 16% of the influent nitrogen for the denitrification/nitrification-MBR plant. The lowest N<sub>2</sub>O emission (EF = 0.5% of the influent nitrogen) was obtained in the biological phosphorus removing bed-MBR plant likely due to an improvement in biological performances exerted by the co-presence of both suspended and attached biomass. The influent C/N ratio has been identified as a key factor affecting the N<sub>2</sub>O production. Indeed, a decrease of the C/N ratio (from 10 to 2) promoted the increase of N<sub>2</sub>O emissions in both gaseous and dissolved phases, mainly related to a decreased efficiency of the denitrification processes.

**Key words** | global warming, MBR, nitrous oxide emission, wastewater treatment

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## INTRODUCTION

During the last decade, the awareness that wastewater treatment plants (WWTPs) are responsible for greenhouse gas emissions (GHGs) has considerably increased as evident from the large number of research papers published on the topic in the last 10 years (Kampschreur *et al.* 2009; Ahn *et al.* 2010; Stenström *et al.* 2014; Ni & Yuan 2015; Mannina *et al.* 2017a). Among the emitted GHGs from WWTPs, N<sub>2</sub>O has been identified of having the major interest. Despite the amount of N<sub>2</sub>O emitted from WWTPs is considerably lower than CO<sub>2</sub> or CH<sub>4</sub>, the major interest on its emission from WWTPs is due to its high global warming potential (GWP), 298 times higher than that of CO<sub>2</sub> for a 100-year time scale (IPCC 2007). Several studies have been performed with the aim to better understand the core pathways of N<sub>2</sub>O formation (Kampschreur *et al.* 2009; Yu & Chandran 2010; Yu *et al.* 2010; Chandran *et al.* 2011; Law *et al.* 2012; Wunderlin *et al.* 2012; Yu *et al.* 2018), most of the studies reported in

literature are related to conventional activated sludge systems (CASs). The acquired knowledge may not be transferred into innovative systems such as membrane bioreactors (MBRs). On the other hand, MBR systems have attracted increasing attention in the last few years, due to several advantages compared to conventional processes (Di Trapani *et al.* 2011a, 2011b). More specifically, MBR systems provide high effluent quality, small footprint and moderate sludge production rates compared to CAS (Stephenson *et al.* 2000).

The specific peculiarities of MBRs (biomass selection; absence of secondary clarifier which can contribute in N<sub>2</sub>O production; intensive aeration for fouling mitigation in membrane compartment which can promote N<sub>2</sub>O stripping, etc.) may hamper a direct transferability of the results derived for CAS systems (Nuansawan *et al.* 2016). Therefore, there is an imperative need to increase the knowledge on N<sub>2</sub>O emission from MBRs, through ad-hoc

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experimental and mathematical modelling activities, since only a few studies have been reported so far (Mannina et al. 2016a, 2016b).

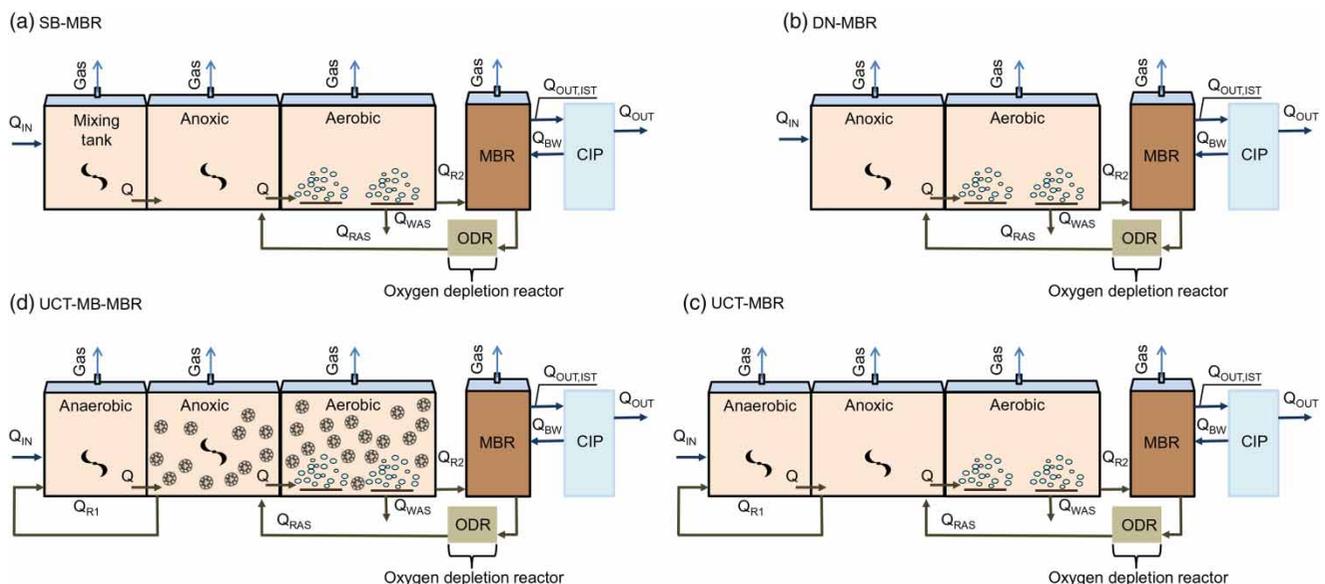
The main goal of this study is to investigate GHG production/emission from MBRs, referring in particular to  $N_2O$ . A detailed investigation where the occurring N transformations are discussed is available in the literature (Mannina et al. 2016a, 2016b; Mannina 2017; Mannina et al. 2017a, 2017b, 2018). Here, different pilot plant configurations, wastewater characteristics and operational conditions have been analysed in a two-year experimental survey (PRIN2012 2012). The goal was to build-up a wide experimental database related to the MBR with the final aim to gain insights and allow, as a further step, the application of mathematical models.

## MATERIAL AND METHODS

Four different pilot plants were investigated during the experimental period (namely, SB-MBR, DN-MBR, UCT-MBR and UCT-IFAS-MBR) (Figure 1). The pilot plants were equipped with specific funnel shape covers that guaranteed gas accumulation in the headspace, to capture the produced gas by sampling. The experimental period had a duration of almost two years and was aimed at investigating the influence of operational variables (SRT, C/N ratio and HRT-SRT), influent features (municipal and industrial wastewater) on  $N_2O$  production and emission (Mannina et al. 2017a, 2017b).

Briefly, pilot plant N.1, referred to as SB-MBR, was designed according to a pre-denitrification scheme and was operated in a sequential feeding mode. It consisted of two in-series reactors: one anoxic (volume: 45 L volume) and one aerobic (volume: 224 L), followed by an MBR compartment (volume: 50 L). The experimental campaign was divided into six phases, each characterized by a different salt concentration in the feeding wastewater. In detail, the salt concentration was gradually increased from 0 to 10 g  $NaCl L^{-1}$  (Phase I: no salt addition; Phase II: 2 g  $NaCl L^{-1}$ ; Phase III: 4 g  $NaCl L^{-1}$ ; Phase IV: 6 g  $NaCl L^{-1}$ ; Phase V: 8 g  $NaCl L^{-1}$ ; Phase VI: 10 g  $NaCl L^{-1}$ ). This campaign was aimed at assessing the system response in terms of  $N_2O$  production when subjected to salt increase in the feeding wastewater, also evaluating the biomass acclimation to a gradual salinity increase in the feeding wastewater. This experimental period was also propaedeutic to the experimental period carried out on pilot plant N.2 (see below), for the treatment of industrial wastewater characterized by the simultaneous presence of salt and hydrocarbons (ship-board slops).

Pilot plant N.2, referred to as DN-MBR, consisted of the same reactors described in pilot plant N.1 with the difference that influent flow rate was fed in continuous mode to the pilot plant. The experimental gathering campaign had a duration of 90 days and was divided in two phases, characterized by different features of the inlet wastewater. In detail, Phase I was characterized by an increasing salinity of the influent (from 10 g  $NaCl L^{-1}$  up to 20 g  $NaCl L^{-1}$ ), while



**Figure 1** | Schematic layout of the investigated pilot plants: SB-MBR (a), pre-denitrification MBR (b), UCT-MBR (c) and UCT-MB-MBR (d).

in Phase II, the inlet wastewater was characterized by constant salinity (20 g NaCl L<sup>-1</sup>) and hydrocarbons dosage (Mannina 2017).

Pilot plant N.3, referred to as UCT-MBR, was characterized by one anaerobic (volume: 62 L), one anoxic (volume: 102 L), and one aerobic (volume: 211 L) in-series reactors, according to the University of Cape Town (UCT) scheme (Ekama *et al.* 1983). The UCT-MBR pilot plant was fed with a mixture of real and synthetic wastewater, the latter characterized by sodium acetate, glycerol, di-potassium hydrogen phosphate and ammonium chloride. The experimental campaign was divided in two phases, each characterized by a different value of the inlet C/N ratio. Phase I, C/N of 10 (duration: 41 days), Phase II, C/N of 5 (duration: 39 days).

Pilot plant N.4, referred to as IFAS-UCT-MBR, was characterized by one anaerobic (volume: 62 L), one anoxic (volume: 102 L), and one aerobic (volume: 211 L) in-series reactors, according to the UCT scheme (Ekama *et al.* 1983). The pilot plant was realized according to the integrated fixed-film activated sludge (IFAS) moving bed biofilm reactor (MBBR) based MBR configuration, with the presence of both activated sludge and biofilm. The suspended plastic carriers for biofilm growth have been added to the anoxic and the aerobic reactors, with filling fraction of 15 and 40%, corresponding to specific area of 75 and 200 m<sup>2</sup> m<sup>-3</sup>, respectively. The experimental campaign has a duration of 340 days and was aimed at investigating the influence of operational variables (namely, SRT, C/N ratio and HRT-SRT) on N<sub>2</sub>O production and emission. The pilot plant was fed with a mixture of real and synthetic wastewater, in order to meet the desired C/N value of the inlet wastewater.

Furthermore, it has to be stressed that in pilot plants N.1 and N.2 the solid-liquid separation was achieved by means of hollow fiber ultrafiltration (UF) module Zenon Zeeweed, ZW 10 (specific area of the filtration module equal to 0.98 m<sup>2</sup> and nominal porosity equal to 0.04 μm). Conversely, pilot plants N.3 and N.4 were equipped with the UF module Koch PURON® 3 bundle with specific area equal to 1.40 m<sup>2</sup> and nominal porosity of 0.03 μm.

Further details on pilot plants description, as well as on experimental campaigns, the reader is addressed to literature (Mannina *et al.* 2016a, 2016b; Mannina 2017; Mannina *et al.* 2017a, 2017b, 2018).

Dissolved and gaseous N<sub>2</sub>O concentrations were measured in each reactor and in the permeate by using a gas chromatograph (Thermo Scientific™ TRACE GC) equipped with an electron capture detector (ECD).

Furthermore, the N<sub>2</sub>O-N fluxes (gN<sub>2</sub>O-N m<sup>-2</sup> h<sup>-1</sup>) from all reactors were quantified by measuring the gas flow rates, Q<sub>GAS</sub> (L min<sup>-1</sup>) (Mannina *et al.* 2017a, 2017b).

Briefly, Q<sub>GAS</sub> was evaluated indirectly, through the following expression (1):

$$Q_{gas} = v_{gas} \cdot A \quad (1)$$

where A represents the outlet section (m<sup>2</sup>) and v<sub>gas</sub> (m s<sup>-1</sup>) is the gas velocity, measured by using a TMA-21HW - Hot Wire anemometer. The gas flow rate is further converted in nitrous oxide flux by taking into account the N<sub>2</sub>O concentration measured in the Q<sub>GAS</sub>.

Gas samples were withdrawn by means of commercial syringes and transferred into glass vials (e.g. LABCO Exetainer, 738 model) where the vacuum was previously created. In order to guarantee the atmospheric pressure inside the vials, the ratio between the volume of the gas sample (inserted inside the vial) and the volume of the vial was no less than 1.25 (e.g. 15 mL of sample in a 12 mL vial). Three replicates were performed for each grab sample. The N<sub>2</sub>O-N concentration was then calculated as the average value among the three replicates.

Dissolved gas sampling was conducted on the basis of the head space gas method derived from Kimochi *et al.* (1998). In detail, 70 mL of supernatant (after 5 min of centrifugation at 8,000 rpm) were sealed into 125 mL glass bottles. To prevent any biological reaction, 1 mL of 2N H<sub>2</sub>SO<sub>4</sub> was added. After 24 h of gentle stirring, the bottles were left for 1 h without moving. Thereafter, the gas accumulated in the headspace of the bottles was collected similarly to the gas sampling procedure.

For each compartment, the evaluation of the N<sub>2</sub>O-N emission factors, expressed as the percentage of N<sub>2</sub>O-N emitted compared to the inlet nitrogen loading rates, was conducted by means of the following Equation (2), derived by Tsuneda *et al.* (2005):

$$EF_{N_2O} = \frac{N_2O - N_{Gas}/HRT_{h,s} + N_2O - N_{Dissolved}/HRT}{TN_{IN}/HRT} \quad (2)$$

where EF<sub>N<sub>2</sub>O</sub> is the emission factor (EF), N<sub>2</sub>O-NGas [mg N<sub>2</sub>O-N L<sup>-1</sup>] is the nitrous dioxide in the gaseous phase, N<sub>2</sub>O-NDissolved [mg N<sub>2</sub>O-N L<sup>-1</sup>] is the nitrous dioxide in the liquid phase, TN<sub>IN</sub> [mg TN L<sup>-1</sup>] is the pilot plant influent total nitrogen concentration, HRT [h] is the hydraulic retention time of the investigated pilot plant while HRT<sub>h,s</sub> [h] represents the headspace retention time of the analysed tank (e.g. the HRT<sub>h,s</sub> of the aerobic reactor corresponds to

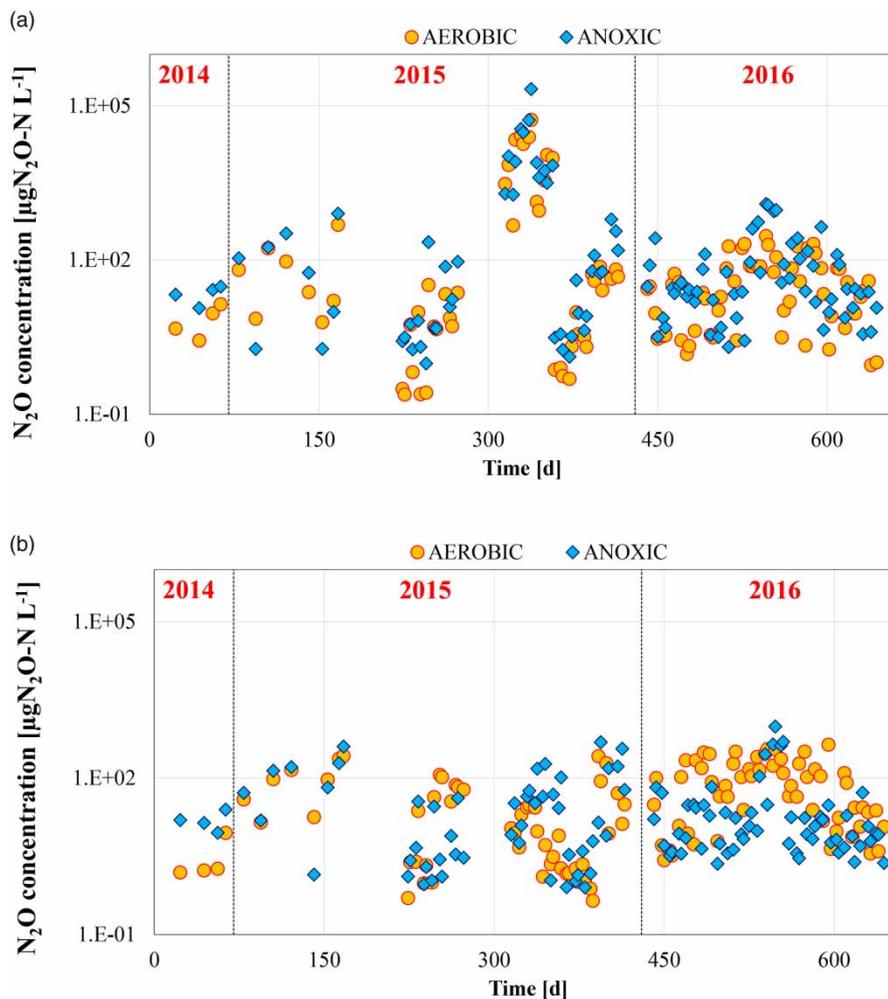
the head space volume of the aerobic reactor divided for the air flow rate evaluated in accordance to Equation (1)).

## RESULTS AND DISCUSSION

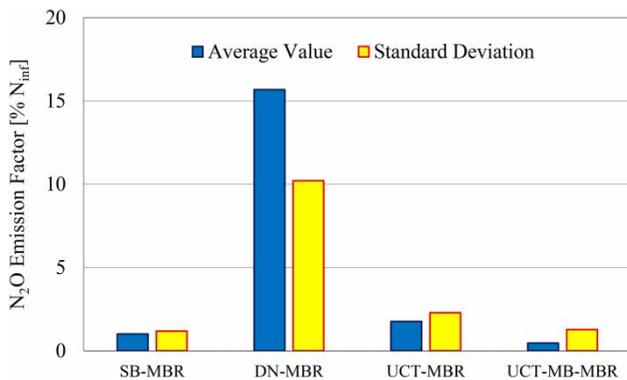
By varying the experimental layout, as noticeable from Figure 1, an intensive campaign aimed at evaluating the nitrous oxide production has been carried out over almost two years. Such effort has been done in order to create a dataset of information useful to provide insights to better understand the nitrous oxide production and emission mechanisms. It is worth noticing that the extreme variability of  $N_2O$  required such a long investigation period; as an example, in Figure 2 the  $N_2O$  gaseous and dissolved concentration measured in the anoxic and aerobic reactors of the aforementioned pilot plants are depicted.

In detail, headspace data are representative of the  $N_2O$  concentration confined between the liquid surface and the funnel shape cover of the reactors, while the dissolved concentration data are representative of the  $N_2O$  present in the liquid bulk of each reactor. It is important to highlight that data depicted in Figure 2 are representative of the reactors, aerobic and anoxic, where the core part of the nitrogen transformation process occur. Data collected over almost two years underline the huge variability of  $N_2O$  concentration measured; the nitrous oxide concentrations ranged within seven orders of magnitude (from  $10^{-1} \mu\text{g } N_2O\text{-N } L^{-1}$  up to  $10^5 \mu\text{g } N_2O\text{-N } L^{-1}$ ).

Such extreme variability in  $N_2O$  concentrations also resulted in a wide range of EF measured during the experimentation. In Figure 3, the average value of emission factors measured for each experimental layout are depicted with the standard deviation.



**Figure 2** | Nitrous oxide concentration measured in the head space (a) and in the liquid bulk (b) of aerobic and anoxic reactors over the experimental campaign.



**Figure 3** | Nitrous oxide average emission factor measured for each experimental layout. The bars report minimum and maximum value for each configuration.

Data depicted in Figure 3 highlight the influence exerted by the specific process scheme on the nitrous oxide emission. In detail, the DN-MBR process showed the highest EF (16% of influent nitrogen on average) as well as the highest standard deviation (close to 10%). It is worth noting that the influent wastewater composition also played a significant role in increasing the N<sub>2</sub>O emission. The DN-MBR scheme treated an influent wastewater composed also by salt and diesel fuel (Mannina *et al.* 2016b). The co-presence of salt and hydrocarbon represented a disturbance factor that, affecting the metabolic activity of biomass, increased the N<sub>2</sub>O production and thus the emission. The role played by the salinity is also noticeable during the operation as SB-MBR configuration. The stepwise salinity increase resulted in a moderate EF (mean EF measured during SB-MBR period resulted equal to 1% of influent nitrogen). The mean dissolved concentration of nitrous oxide during SB-MBR operation ranged from 1.5 µg N<sub>2</sub>O-N L<sup>-1</sup> to 38.5 µg N<sub>2</sub>O-N L<sup>-1</sup>. A similar concentration was measured in the head space during the DN-MBR configuration. During the SB-MBR period, the maximum nitrous oxide concentrations, dissolved and in the head space, were measured constantly in the anoxic reactor. Such an occurrence highlights that the main N<sub>2</sub>O production pathway during SB-MBR configuration was the anoxic heterotrophic denitrification.

With regard to the UCT-MBR and IFAS-UCT-MBR configuration, the scarcity of carbon availability imposed during the lowest values of C/N ratio resulted in an increase of N<sub>2</sub>O emission likely due to a limitation of denitrification process and to the inhibition of the complete nitrification (Mannina *et al.* 2017b, 2018). A significant NO<sub>2</sub> accumulation (till to the average value of 30 mg L<sup>-1</sup>) in the aerobic reactor occurred in the IFAS-UCT-MBR configuration at C/N equal to 2, resulting in incomplete nitrification due to nitrite oxidizing bacteria inhibition (Mannina *et al.* 2017c). Conversely, a

slight accumulation of NO<sub>2</sub> occurred during C/N equal to 5 for the UCT-MBR configuration (1.2 mg L<sup>-1</sup>) (Mannina *et al.* 2016c).

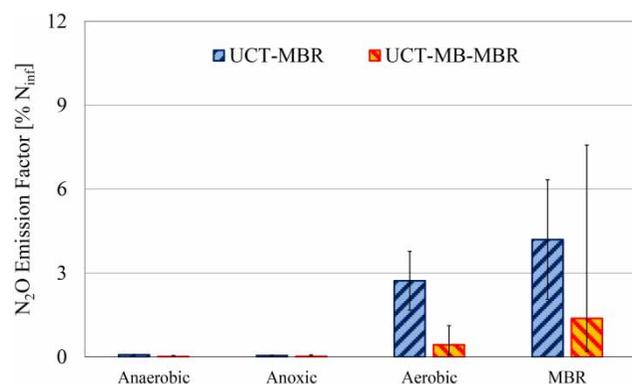
To summarize, the configuration that yielded the lowest EF was the UCT-MB-MBR that was featured by a mean emission equal to 0.5% of influent nitrogen. Actually, the operational condition influenced the emission also during this period. As an example, when an SRT = 30 d was imposed to the pilot plant, the mean EF resulted equal to 7.57%.

In order to also describe the role played by each reactor in contributing to the total emission, in Figure 4 is depicted a comparison of mean EF assessed for each reactor during UCT-MBR and IFAS-UCT-MBR configuration. Data in Figure 4 highlight the strong reduction in EF during the IFAS-UCT-MBR layout.

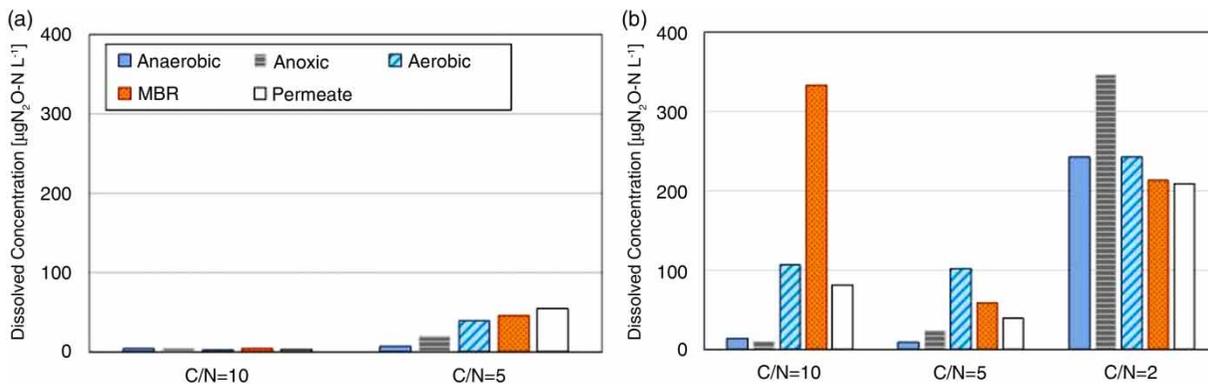
Such a result is likely due to an improvement in biological performances exerted by the co presence of both suspended and attached biomass. The biofilm presence improved the nitrogen removal efficiency thus leading to a lower N<sub>2</sub>O emission. It is worth noticing that the highest emissions were measured, during operation of both layout, in the aerated reactors (aerobic reactor and MBR compartment), thus confirming the significant role played by the aeration devices in enhancing the N<sub>2</sub>O stripping favouring the emission.

In order to better discuss the influence played by the operational conditions as well as by the layout configurations on the nitrous oxide production, in Figure 5 a comparison of mean N<sub>2</sub>O dissolved concentration measured in the biological reactors during C/N period of UCT-MBR and IFAS-UCT-MBR layout is depicted.

Data reported in Figure 5 allow noticing the increase in N<sub>2</sub>O dissolved concentration measured due to the C/N



**Figure 4** | Comparison of mean EF measured in each biological reactor during the UCT-MBR and IFAS-UCT-MBR layout.



**Figure 5** | Comparison of mean N<sub>2</sub>O dissolved concentration measured in the biological reactors during C/N period of UCT-MBR (a) and IFAS-UCT-MBR (b) layout.

reduction. Such a result that was achieved by using real domestic wastewater is consistent with previous findings (Itokawa *et al.* 2001; Alinsafi *et al.* 2008; Kampschreur *et al.* 2009; Lu & Chandran 2010). In detail during the UCT-MBR layout (Figure 5(a)), the decrease of C/N fed to the pilot plant resulted in a drastic increase of dissolved nitrous oxide concentration. As an example, the mean N<sub>2</sub>O concentration measured in the permeate flow during C/N = 10 period was equal to 9.21 μg N<sub>2</sub>O-N L<sup>-1</sup>. Conversely, when the carbon availability was reduced due to the lower C/N, the mean N<sub>2</sub>O concentration measured in the permeate flow increased up to 95.85 μg N<sub>2</sub>O-N L<sup>-1</sup>, thus 10 times higher than the preceding period.

Also for the IFAS-UCT-MBR configuration (Figure 5(b)), a high emission occurred at the low C/N ratio; however, it is worth noticing that, during the C/N = 2 period, the highest N<sub>2</sub>O dissolved concentration was measured in the anoxic reactor. In detail, during C/N = 10 and C/N = 5 period, the maximum values of N<sub>2</sub>O concentration were measured in the aerobic or in the MBR reactor. When the C/N = 2 was imposed, the absence of organic carbon available for denitrification led to a sharp increase of nitrous oxide production likely in the denitrification zone that reached the average concentration of 347 μg N<sub>2</sub>O-N L<sup>-1</sup>.

The mean N<sub>2</sub>O dissolved concentration measured in the anoxic reactor during C/N = 10 and C/N = 5 were equal to 9.47 μg N<sub>2</sub>O-N L<sup>-1</sup> and 21.45 μg N<sub>2</sub>O-N L<sup>-1</sup>, respectively. Similar considerations can also be done by observing the N<sub>2</sub>O concentration measured in the anaerobic reactor during the C/N = 2 period. Indeed, the mean N<sub>2</sub>O dissolved concentration in the anaerobic increased up to 244.20 μg N<sub>2</sub>O-N L<sup>-1</sup> in the C/N = 2 period, thus more than 10 times higher than the other periods. These results are due to the absence of organic carbon available for the denitrification that resulted also in an increase of

nitrites recycled to the anaerobic reactor. As a consequence, during the C/N = 2 period, the anaerobic reactor resulted in being overloaded by nitrites, thus acting as an anoxic reactor and contributing to the high N<sub>2</sub>O production.

## CONCLUSIONS

In this study, the overall assessment of N<sub>2</sub>O emissions from fully covered MBR pilot plants was performed. Different MBR configurations, operational conditions and influent features were investigated in view of better understanding the key factors mainly influencing the N<sub>2</sub>O emission in MBRs.

The following conclusions can be drawn:

- The aerated reactors (including MBR) represent the main source in terms of N<sub>2</sub>O emission for all the investigated configurations. Therefore, in the MBRs the aeration devices play a key role in enhancing the N<sub>2</sub>O stripping favouring the emission.
- A strong variability of N<sub>2</sub>O produced/emitted from each tank was observed.
- The highest absolute N<sub>2</sub>O emission (16% of the influent nitrogen) was observed when industrial wastewater contaminated with salt and hydrocarbons were treated (in a DN-MBR configuration); therefore, particular attention has to be paid on gases produced from industrial WWTPs.
- Biofilm systems allow an improvement of the nitrogen removal efficiency thus leading to a lower N<sub>2</sub>O emission.
- The influent C/N ratio has a key role in promoting/reducing the N<sub>2</sub>O emission. In a biofilm system (IFAS-UCT-MBR), the reduction of the influent C/N

ratio (from 10 to 2) led to the doubling of the N<sub>2</sub>O dissolved concentration of each tank.

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