

Corrigendum to "Thermodynamics of soluble fission products cesium and iodine in the Molten Salt Reactor" [Journal of Nuclear Materials Volume 501 (2018) pages 238-252]

Capelli, E.; Beneš, O.; Konings, R. J.M.

10.1016/j.jnucmat.2021.153156

**Publication date** 

**Document Version** Final published version

Published in Journal of Nuclear Materials

Citation (APA)
Capelli, E., Beneš, O., & Konings, R. J. M. (2021). Corrigendum to "Thermodynamics of soluble fission products cesium and iodine in the Molten Salt Reactor" [Journal of Nuclear Materials Volume 501 (2018) pages 238-252]. *Journal of Nuclear Materials*, 555, Article 153156. https://doi.org/10.1016/j.jnucmat.2021.153156

To cite this publication, please use the final published version (if applicable). Please check the document version above.

Copyright

Other than for strictly personal use, it is not permitted to download, forward or distribute the text or part of it, without the consent of the author(s) and/or copyright holder(s), unless the work is under an open content license such as Creative Commons.

Please contact us and provide details if you believe this document breaches copyrights. We will remove access to the work immediately and investigate your claim.

\$ SEVIER

Contents lists available at ScienceDirect

## Journal of Nuclear Materials

journal homepage: www.elsevier.com/locate/jnucmat



## Corrigendum

## Corrigendum to "Thermodynamics of soluble fission products cesium and iodine in the Molten Salt Reactor" [Journal of Nuclear Materials Volume 501 (2018) pages 238-252]



E. Capelli<sup>a,b,\*</sup>, O. Beneš<sup>c</sup>, R.J.M. Konings<sup>a,c</sup>

- <sup>a</sup> Department of Radiation Science and Technology, Faculty of Applied Sciences, Delft University of Technology, Delft 2629JB, The Netherlands
- <sup>b</sup> Nuclear Research and Consultancy Group (NRG), 1755LE Petten, The Netherlands
- <sup>c</sup> European Commission, Joint Research Centre, 76125 Karlsruhe, Germany

The authors regret to inform that the thermodynamic data  $\Delta_f H^{298}$  of the pure compound Csl(l) was not reported correctly in Table 3. The corresponding enthalpy of formation should read:

 $\Delta_f H^{298} (kJ \cdot mol^{-1}) = -331.912$