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# Characterisation of glass polishing waste samples

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## Abstract

The present work describes the characterisation of the two different polishing waste samples. The samples were analysed with laser particle size analyser, XRD, XRF, TGA, and SEM. In the sample A, a large amount of calcite ( $\text{CaCO}_3$ ) together with silica and aluminosilicates were observed with SEM. In the mineralogy, only  $\text{CaCO}_3$ ,  $\text{CeO}_2$  and  $\text{LnO}_{0.65}\text{F}_{1.7}$  were found. In the sample B, very small amounts of impurities (<2%) were found. The particle size of sample B was decreased compared to its original polishing powder.  $\text{CeO}_2$  and  $\text{LnO}_{0.65}\text{F}_{1.7}$  compounds were found in the XRD analysis.

## Introduction

Ceria is the principal compound in glass polishing powder because of its chemical and mechanical properties. After several cycles of polishing, this powder either enriches with impurities (glass, flocculants etc.) or changes its particle size distribution. Hence, it can't be further used due to poor polishing properties and ends up in landfills. The present work describes the characterisation of the two different polishing waste samples, which will help in designing the subsequent recovery processes.

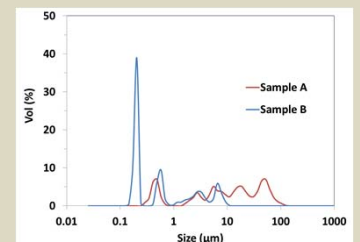
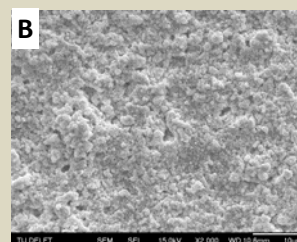
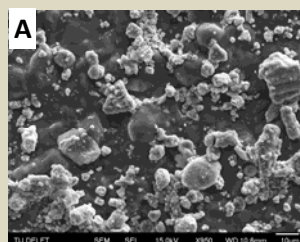
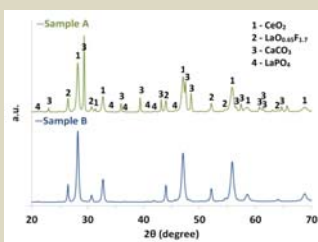
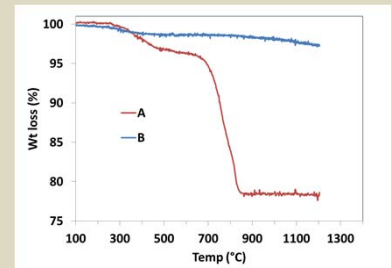
## Methods and Materials

- Two glass polishing waste samples
- XRF – Chemical analysis
- TGA – Thermal analysis
- XRD – Mineralogy
- SEM – Morphology and mineralogy
- LD – Particle size distribution

## Results and Discussion



(wt. %)	A	B
$\text{CeO}_2$	22.1	55.3
$\text{La}_2\text{O}_3$	12.7	29.5
F	3.2	8.3
$\text{SiO}_2$	5.7	0.4
$\text{Al}_2\text{O}_3$	4.3	-
CaO	27.9	0.2
$\text{Fe}_2\text{O}_3$	0.4	0.7
$\text{P}_2\text{O}_5$	0.8	2.2
LOI	21.5	1.7



## Conclusions

- Sample B contains very less impurities where sample A contains high amount of impurities.
- Sample B contains compounds of  $\text{CeO}_2$ ,  $\text{LaO}_{0.65}\text{F}_{1.7}$  and  $\text{LaPO}_4$ .
- Sample A contains compounds of  $\text{CeO}_2$ ,  $\text{CaCO}_3$ ,  $\text{LaO}_{0.65}\text{F}_{1.7}$  and  $\text{LaPO}_4$  together with silica and aluminosilicates
- Sample B is fine material compared to sample A

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